Amide vibrations are delocalized across the hydrophobic interface of a transmembrane helix dimer

Chong Fang[†], Alessandro Senes[‡], Lidia Cristian[‡], William F. DeGrado[‡], and Robin M. Hochstrasser^{†§}

[†]Department of Chemistry, University of Pennsylvania, Philadelphia, PA 19104-6323; and [‡]Department of Biochemistry and Biophysics, University of Pennsylvania School of Medicine, Philadelphia, PA 19104-6059

Contributed by Robin M. Hochstrasser, September 20, 2006

The tertiary interactions between amide-I vibrators on the separate helices of transmembrane helix dimers were probed by ultrafast 2D vibrational photon echo spectroscopy. The 2D IR approach proves to be a useful structural method for the study of membrane-bound structures. The 27-residue human erythrocyte protein Glycophorin A transmembrane peptide sequence: KKITLIIFG79VMAGVIGTILLISWG94IKK was labeled at G79 and G94 with ¹³C=¹⁶O or ¹³C=¹⁸O. The isotopomers and their 50:50 mixtures formed helical dimers in SDS micelles whose 2D IR spectra showed components from homodimers when both helices had either $^{13}\text{C}=^{16}\text{O}$ or $^{13}\text{C}=^{18}\text{O}$ substitution and a heterodimer when one had ¹³C=¹⁶O substitution and the other had ¹³C=¹⁸O substitution. The cross-peaks in the pure heterodimer 2D IR difference spectrum and the splitting of the homodimer peaks in the linear IR spectrum show that the amide-I mode is delocalized across a pair of helices. The excitation exchange coupling in the range 4.3-6.3 cm⁻¹ arises from through-space interactions between amide units on different helices. The angle between the two Glv79 amide-I transition dipoles, estimated at 103° from linear IR spectroscopy and 110° from 2D IR spectroscopy, combined with the coupling led to a structural picture of the hydrophobic interface that is remarkably consistent with results from NMR on helix dimers. The helix crossing angle in SDS is estimated at 45°. Two-dimensional IR spectroscopy also sets limits on the range of geometrical parameters for the helix dimers from an analysis of the coupling constant distribution.

tertiary interaction | vibrational spectra | multidimensional spectroscopy

Two-dimensional IR spectroscopy (1-6) is a promising new method with which to probe structures and their motions in complex systems. Recent applications of this method to biologically related molecules have yielded novel structural and dynamical results not readily obtainable by other methods for small peptides (7, 8), soluble helices (9, 10), membrane-bound helices (11), lipids (12), β -sheets (13, 14), and model secondary structures (15, 16). The 2D IR approach is one that can be immediately applicable to a wide variety of sample types ranging from solutions to solids, including aqueous and lipid environments. The method exposes interactions between spatially nearby vibrational modes by converting three pulse photon echo signals into 2D spectral maps in the IR, analogous to 2D NMR spectra.

The amide-I vibrational bands of polypeptides are highly degenerate because, for each residue, their frequencies are approximately the same and the interactions among them are not strong enough to separately display them as individual, assignable transitions. The combination of multiple isotope selection and 2D IR spectroscopy that spreads the transitions into two dimensions proves to be a useful strategy that can simplify and expose the underlying transitions of the diffuse amide bands and allow their features to be characterized at a residue level. For example the 2D IR spectrum of a water-soluble helix selectively substituted with ^{13}C — ^{16}O and ^{13}C — ^{18}O provided characteristics of pairs of coupled residues and

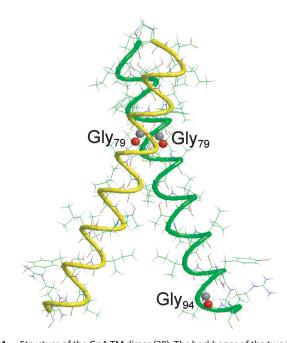


Fig. 1. Structure of the GpA TM dimer (20). The backbones of the two helical segments are yellow and green. The carbonyl atoms of the interfacial Gly_{79} positions and the control Gly_{94} position are represented as spheres.

the visualization of the sequence dependence of the structural dynamics (17).

This paper introduces a comparable approach to investigate tertiary interactions between vibrational modes in Glycophorin A (GpA), a transmembrane (TM) dimer of interacting helices (Fig. 1). The GpA TM helix provides an excellent prototype for the study of TM helix association, a key event in the folding of membrane proteins (18). The dimer also is an appropriate model for these investigations, because it is stable in a variety of micelles, including SDS (19), and a solution NMR structure is available (20). There are two Gly residues at the dimer interface that allow an intimate contact of the main chain, a feature frequently observed in TM helical interfaces (21, 22). Because vibrational coupling is sensitive to distance, the proximity of the backbones is advantageous. GpA also was selected because verification of the applicability of the 2D IR method to membrane protein aggregates is of great interest and because there is a chronic paucity of structural information for this

Author contributions: W.F.D., A.S., and R.M.H. designed research; C.F., A.S., and L.C. performed research; C.F. analyzed data; and C.F. and R.M.H. wrote the paper.

The authors declare no conflict of interest.

Abbreviations: GpA, Glycophorin A; TM, transmembrane; TFA, trifluoroacetic acid.

[§]To whom correspondence should be addressed. E-mail: hochstra@sas.upenn.edu.

^{© 2006} by The National Academy of Sciences of the USA

class of proteins. The angular information that is obtainable with the present method can be used to help derive molecular models. This information is especially useful because the simple topology generally adopted by membrane proteins (bundles of roughly parallel α -helices) is such that computational procedures can be effectively assisted by a relatively small number of spatial constraints (23–28), particularly when a regular helix conformation can be assumed, which is likely to be the case for oligomeric complexes of single helices such as GpA.

The Labeling Notation

Four peptides were prepared in the study of the GpA dimer interface by linear and 2D IR spectroscopy. The unlabeled peptide, denoted as G_{79} , had a deuterium label at the α -carbon of Gly₇₉, which we presumed has no measurable effect on the amide-I transitions. The ¹³C=¹⁶O-labeled peptides were denoted as either G_{79}^* or G_{94}^* , and the ¹³C=¹⁸O-labeled peptide was denoted as G_{79}^{**} . The samples of pure G_{79}^* and G_{79}^{**} yield in the micelle either $^{13}\text{C}=^{16}\text{O}$ or $^{13}\text{C}=^{18}\text{O}$ homodimers, denoted as G_{79}^*/G_{79}^* and G_{79}^{**}/G_{79}^{**} , respectively. A 50:50 mixture of G_{79}^{*} and G_{79}^{**} , denoted as $G_{79}^* + G_{79}^{**}$, yields one part of each of the homodimers in addition to one each of the heterodimers denoted as G_{79}^{**}/G_{79}^* and G_{79}^*/G_{79}^{**} . Peptide G_{94}^* , which has ${}^{13}C = {}^{16}O$ at residue 94 was used to obtain G_{94}^*/G_{94}^* and $G_{94}^*+G_{79}^{**}$, with the latter yielding homodimers and heterodimers in the micelle. The mixture $G_{79} + G_{79}^{***}$ of unlabeled and ¹³C=¹⁸O-labeled GpA was used as a control. The typical GpA TM helix dimer structure (20) is illustrated in Fig. 1.

Results

Linear IR Spectroscopy. The FTIR results are shown in Fig. 2 a and b with the isotopomer region expanded and each spectrum normalized to the optical density of the main TM helical band of G_{79} . The main amide-I' bands are in the region 1,630–1,660 cm⁻¹, and the band at 1,674 cm⁻¹ is trifluoroacetic acid (TFA). The lowfrequency $(1,630 \text{ cm}^{-1})$ part of the main helix band is dominated by a secondary structure that is not imbedded in the micelle, with the shift to lower wave number arising from hydration of the amide groups (1, 29). Neglecting TFA, the amide-I' spectrum of G_{79} can be fitted to two broad transitions, one at ≈1,635 cm⁻¹ that is associated with misfolded residues or those outside the micelle and another that is \approx 2.3 times stronger at \approx 1,655 cm⁻¹ and corresponds to the structure in the micelle. The ${}^{13}\text{C}={}^{16}\text{O}$ transition of G_{79}^* appears at \approx 1,613 cm⁻¹. The ¹³C=¹⁶O natural abundance transitions of G_{79} at $\approx 1,615$ cm⁻¹ have a much reduced intensity due to the large spread of transition frequencies and relaxation times that occur within the random residue distribution. The line shape of the G_{79}^{***} isotopomer is noticeably asymmetric and broader than the other transitions, indicating that it corresponds to at least two underlying spectral components. It will be seen that this splitting is the result of an exciton interaction. This asymmetric ¹³C=¹⁸O region of G_{79}^{**} after the subtraction of G_{79} to remove the background was least-squares-fitted to two Gaussian bands at 1,589.5 \pm 0.8 cm⁻¹ and 1,598.1 \pm 0.8 cm⁻¹, which are separated by 8.6 \pm 1.6 cm^{-1} , as shown in Fig. 2c. Both of the widths of the two underlying bands are 14.0 ± 1.5 cm⁻¹. The intensity ratio of the low-frequency to high-frequency component in G_{79}^{**} is 1.6 \pm 0.1. A comparable asymmetry is observed for the isotope region of G_{79}^* , but it is not as convincingly analyzed because of the underlying broad and symmetric ¹³C=¹⁶O natural abundance contribution at about the same frequency. The absorption cross-section of ${}^{13}C = {}^{16}O$ is ≈ 1.3 times larger than ¹³C=¹⁸O because the ¹³C=¹⁶O states are coupled more strongly to the neighboring ¹²C=¹⁶O states.

Two-Dimensional IR Spectroscopy. The expectations for a mixed $^{13}\text{C} = ^{16}\text{O}$ and $^{13}\text{C} = ^{18}\text{O}$ isotopomer are that the isotopically substituted amide transitions of a helix are split by the difference in the isotope shifts for the $^{13}\text{C} = ^{16}\text{O}$ ($\approx 38-43$ cm $^{-1}$) and $^{13}\text{C} = ^{18}\text{O}$ ($\approx 62-67$ cm $^{-1}$) substitutions (17). The estimated (see

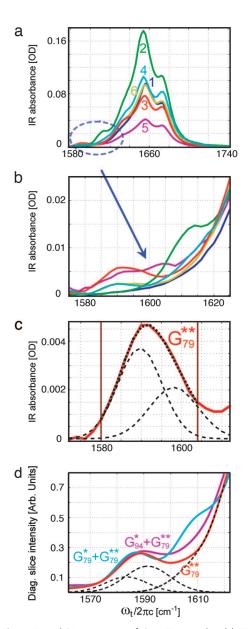


Fig. 2. Linear IR and 2D IR spectra of GpA TM samples. (a) FTIR spectra (pathlength, 25 μ m) of homodimers G_{79} (spectrum 1), G_{79}^* (spectrum 2), and G_{79}^* (spectrum 3) and heterodimers G_{79}^* + G_{79}^* (spectrum 4), G_{54}^* + G_{79}^* (spectrum 5), and G_{79} + G_{79}^* (spectrum 6) in 5% SDS. (b) Magnification of the ^{13}C — ^{18}O and ^{13}C — ^{16}O bands circled in a. (c) The fitting of the ^{13}C — ^{18}O mode of G_{79}^* in FTIR showing the two underlying components as dashed black lines and the sum as a dotted line. (d) The diagonal slices of the rephasing magnitude spectra (in Fig. 3) and the fitted components in G_{79}^{*} (dark lines).

below) zero-order frequency separation in the present case is $\approx 21~\rm cm^{-1}$. If the selected modes are coupled, cross-peaks should appear in the 2D IR spectra. Each diagonal 2D IR band consists of two components having opposite signs corresponding to the $v=0{\rightarrow}1$ and $v=1{\rightarrow}2$ transition regions, the latter being shifted to lower frequency by the diagonal anharmonicity. Each crosspeak also will consist of two overlapping bands having opposite signs and separated by the mixed mode anharmonicity. The results can be interpreted in terms of a two-state quantum-mechanical description of the selected pair of amide modes when they are both significantly frequency-down-shifted from the dominant ^{12}C — ^{16}O bands of the peptide.

The cross-peaks also are seen in the absolute magnitudes of the

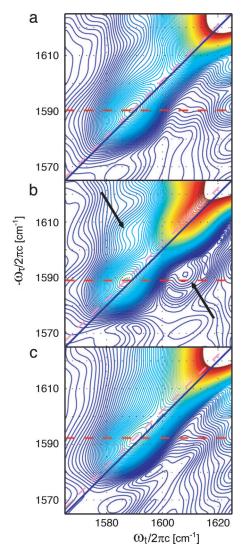


Fig. 3. The rephasing magnitude spectra of homodimer $G_{79}^{**}(a)$, heterodimer $G_{79}^* + G_{79}^{**}$ (b), and control heterodimer $G_{94}^* + G_{79}^{**}$ (c). The horizontal line indicates the ¹³C=¹⁸O peak position, and the diagonal line is a slice of the spectrum shown in Fig. 2d. The arrows point to the cross-peaks.

2D IR spectra for G_{79}^{**} and $G_{79}^{*} + G_{79}^{**}$ (Fig. 3 a and b). Fig. 2d shows traces of these spectra along the diagonals through the ¹³C=¹⁸O peak centers. In this representation, the spectrum is positive everywhere, whereas in the absorptive 2D IR spectrum, the interference alters the cross-peak shapes to make them less obvious. The control sample $G_{94}^* + G_{79}^{**}$ (Fig. 3c) shows no cross-peaks. The 2D correlation spectra of various GpA samples are shown in Fig. 4. The echo signals are dominated by the elliptically shaped diagonal peaks corresponding to the $v = 0 \rightarrow 1$ (red) and $v = 1 \rightarrow 2$ (blue) transitions. The regions where the ¹³C=¹⁶O and ¹³C=¹⁸O diagonal transitions are expected are highlighted as rectangles. The G_{79} sample (Fig. 4a) shows only a weak natural abundance peak, whereas G_{79}^* and G_{79}^{**} show the spectra of the homodimers G_{79}^*/G_{79}^* (Fig. 4b) and G_{79}^{**}/G_{79}^{**} (Fig. 4c). These spectra consist only of diagonal features in the isotope region. The sample $G_{79}^* + G_{79}^{**}$ (Fig. 4d), which incorporates the heterodimer spectrum, shows features that are not on the diagonal of the 2D IR spectrum. The positive hump at the higherfrequency side of the ¹³C=¹⁸O mode and the negative dip to the lower-frequency side of the ¹³C=¹⁶O mode are attributed to cross-peaks, which locate at the corners of the square that they form along with the two diagonal isotopomer levels at 1,589.3 \pm 0.6 cm⁻¹ and 1,613.8 \pm 0.6 cm⁻¹. The control sample $G_{94}^* + G_{79}^{**}$ (Fig. 4f) does not show these off-diagonal features. In the difference spectrum $(G_{79}^* + G_{79}^{**}) - 0.25G_{79}^{**}$, where the G_{79}^{**}/G_{79}^{**} contribution is removed and the 13C=18O spectral region arises from the pure heterodimers G_{79}^{**}/G_{79}^* and G_{79}^*/G_{79}^{**} (see Supporting Text, which is published as supporting information on the PNAS web site), the cross-peaks become more prominent both above and below the diagonal (Fig. 4e).

The prominent cross-peak in the spectral region of $\omega_t \approx$ 1,615–1,635 cm⁻¹ below the diagonal appearing whenever there is a ${}^{13}\text{C}={}^{18}\text{O}$ substitution (Figs. 4 c–f) signifies the coupling of the ¹³C=¹⁸O modes to the unlabeled exciton band states at higher frequencies: Both tertiary and secondary interactions can contribute.

Angular Measurements. The relative magnitudes of spectra of $(G_{79}^* + G_{79}^{**}) - 0.25G_{79}^{**}$ in different polarizations are related to $P_2(\cos\theta_{12})$, where θ_{12} is the angle between the transition dipole moments on the two amide groups (30). One measure is the ratio of the \(\lambda xxxx\rangle\) signal amplitude of the \(^{13}C=^{18}O\) diagonal to that of the cross-peak (2.0 \pm 0.18). Another is the ratio of $\langle xxxx \rangle / \langle xyyx \rangle$ at the cross-peak (see Supporting Text), which has an undetermined sign from $P_2 = P_2(\cos\theta_{12})$. For the choice $\langle xyyx \rangle > 0$, we find 3.9 ± 0.3 from least-squares fitting, and for $\langle xyyx \rangle < 0$ we obtain the ratio -5.0 ± 0.5 . In the rephasing magnitude spectra the signal in the cross-peak region (Fig. 3b) is 3.34 times stronger in the $\langle xxxx \rangle$ than in the $\langle xyyx \rangle$ spectrum. These experimental observations can be combined to yield estimates for the angle θ_{12} within certain limitations (31). The shape of the G_{79}^{***} spectrum in the $^{13}\text{C}=^{18}\text{O}$ region (data not shown) distinctly depends on polarization, directly implicating a dimer with split symmetric and antisymmetric components.

Anharmonicities. Fig. 5a shows the trace of the 2D IR difference correlation spectrum of $(G_{79}^* + G_{79}^{**}) - 0.25G_{79}^{**}$ along the ω_t axis, averaged over a Gaussian distribution (12 cm⁻¹ full width at half maximum) on ω_{τ} centered at $\omega_{\tau} = 1,590 \text{ cm}^{-1}$, the location of the ¹³C=¹⁸O transition. Fig. 5b compares correlation spectra of $G_{79}^* + G_{79}^{**}$ and $G_{94}^* + G_{79}^{**}$ as traces without any subtractions. All of the features except the cross-peak appear in both spectra. The diagonal peaks in these traces were least-squares-fitted to pairs of positive $(v = 1 \rightarrow 2)$ and negative $(v = 0 \rightarrow 1)$ peaks separated by a frequency gap equal to the diagonal anharmonicity. The off-diagonal anharmonicity from the fits was $\Delta_{AB} = 3.8 \pm 0.6$ cm⁻¹, implying that the frequency of the amide-I combination band of the two isotopomers is smaller than the sum of their frequencies. Similar dynamical parameters were assumed for the cross-peak and the ¹³C=¹⁸O diagonal peak. The pure heterodimers G_{79}^{**}/G_{79}^{*} and G_{79}^{*}/G_{79}^{**} have a diagonal anharmonicity of the ¹³C=¹⁸O amide-I' mode of $\Delta = 11.6 \pm 0.5$ cm⁻¹, and a $v = 0 \rightarrow 1$ transition at 1,589.5 \pm 0.3 cm⁻¹.

The signals in the cross-peak region discussed above are absent in the sample $G_{94}^* + G_{79}^{**}$ (Fig. 4f), consistent with residues 79 and 94 (see Fig. 1) having negligible interaction. Comparing Fig. 4 d with f or Fig. 3 b with c, the still obvious ${}^{13}C = {}^{16}O$ region of the 2D IR spectrum of $G_{94}^* + G_{79}^{**}$ (compared with Fig. 4 a and c) is more extended along the diagonal (see also Fig. 2b, spectrum 5), apparently forming two peaks with the lower frequency one on the edge of the ¹³C=¹⁸O peak. This result suggests that the structure near Gly₉₄ has a more disordered distribution of amide modes.

Discussion

New structural and dynamical parameters are obtained from the FTIR and 2D IR spectra of both the homodimers and heterodimers of ¹³C=¹⁶O and ¹³C=¹⁸O substituted TM GpA helices. To describe the homodimer spectra, G_{79}^*/G_{79}^* and G_{79}^{**}/G_{79}^* G_{79}^{***} are each treated as pairs of coupled degenerate states considered to be isolated from the remaining ¹²C=¹⁶O amide modes. The heterodimers, treated as pairs of nondegenerate

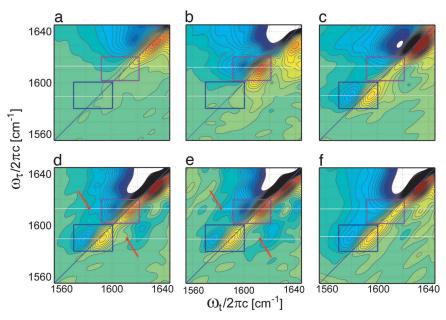


Fig. 4. The 2D correlation spectra of GpA TM homodimers G_{79} (a), G_{79}^* (b), and G_{79}^* (c); heterodimers $G_{79}^* + G_{79}^*$ (d) and $G_{79}^* + G_{79}^*$ (e); and control sample $G_{94}^* + G_{79}^*$ (f) in 5% SDS. The $^{13}C_{13}^{-18}O$ and $^{13}C_{13}^{-18}O$ isotopomer diagonal regions are highlighted by rectangular boxes, and the arrows point to the cross-peaks.

states split by the different isotope shifts, should exhibit crosspeaks that evidence the coupling between the two helices. The exciton model provides a consistent picture of all of the results.

The FTIR spectra of the homodimers are explained by the simple two-level model where the degenerate isotopic levels are split by the tertiary interaction into a symmetric and antisymmetric pair, each of which is completely delocalized over the two helices. The splitting between these states is twice the coupling constant β_{AB} . The two transitions in the $^{13}C=^{18}O$ region of G_{79}^{**} therefore indicate that $2|\beta_{AB}|=8.6\pm1.6$ cm $^{-1}$. For such a two-level model, the ratio of the integrated strength of the lower to the higher frequency component is $(1 \mp \cos\theta_{12})/(1 \pm \cos\theta_{12})$: The upper (lower) sign is used when the coupling is positive (negative). The analysis based on the observed ratio of 1.6 ± 0.1 yields $\theta_{12}=103.0\pm2.0^{\circ}$ for a coupling of +4.3 cm $^{-1}$ or $\theta_{12}=77.0\pm2.0^{\circ}$ for a coupling of -4.3 cm $^{-1}$. The amplitudes and polarizations of the homo- and heterodimer 2D IR spectra are consistent with this two-level model.

The FTIR spectra of the heterodimers also display two transitions, which, according to an exciton model, are expected at

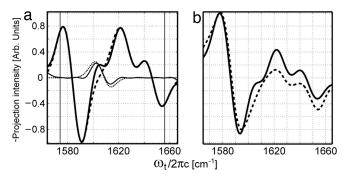


Fig. 5. Traces of 2D IR correlation spectra. (a) Least-squares fit to the 12 cm $^{-1}$ -wide trace along the ω_t axis of the heterodimer $(G_{79}^*+G_{79}^{**})-0.25G_{79}^{***}$ as described in the text (thick solid line), the fit omitting the cross-peak (thick dashed line), the difference between the solid and the dashed line (thin solid line), and the cross-peak response from least-squares-fitting the whole spectral region between the two vertical lines (thin dotted line). (b) Comparison between the $^{13}C_{-}^{-18}O$ trace of $G_{79}^*+G_{79}^{***}$ (solid line) and the control sample $G_{94}^*+G_{79}^{***}$ (dashed line).

frequencies ω_+ and ω_- given by: $\omega_\pm = 1/2[(\omega_A^{(16)} + \omega_B^{(18)} \pm (\delta_{AB}^2 + 4\beta_{AB}^2)^{1/2}]$, where the frequencies for the $^{13}C_-^{16}O$ and $^{13}C_-^{18}O$ substitutions on helix A are labeled $\omega_A^{(16)}$ and $\omega_A^{(18)}$ and those for B are $\omega_B^{(16)}$ and $\omega_B^{(18)}$, and $\delta_{AB} = \omega_A^{(16)} - \omega_B^{(18)}$. If the two amide units are related by a twofold axis of symmetry, then $\omega_A^{(16)} - \omega_B^{(18)} = \omega_B^{(16)} - \omega_A^{(18)}$, and the heterodimers G_{79}^{**}/G_{79}^{**} and G_{79}^{*}/G_{79}^{**} become equivalent. Such symmetry also is required for the homodimers to be considered as coupled degenerate states. The first term in ω_\pm is the average frequency of the two isotopically substituted amide modes, which is given from experiment as $\approx 1,603$ cm⁻¹. The second term, which is measured by the difference $(\omega_+ - \omega_-)$, is found to be ≈ 24.5 cm⁻¹. Our results show that the energy difference between the two isotopomer modes of 21.0 ± 0.5 cm⁻¹ increases to 24.5 ± 0.5 cm⁻¹ in the heterodimer sample, which provides another independent measure of the coupling constant magnitude as $|\beta_{AB}| = 6.3 \pm 0.5$ cm⁻¹.

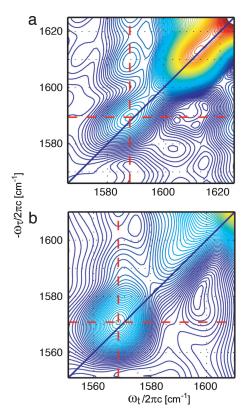
The 2D IR cross-peaks provide the strongest evidence of the delocalization of the vibrational excitations across the two helices. These cross-peaks are a direct measure of the anharmonic coupling Δ_{AB} of the modes, and, because there is no source of anharmonicity between two amide units on different helices except for their electrostatic interaction, the anharmonicity must be related to this coupling. The value of Δ_{AB} can be estimated from a quantum mechanical model (3) that requires as input the diagonal anharmonicity Δ and the separation between the two isotopomer states in the absence of coupling between them. The model yields $\Delta_{AB}=4\Delta\beta_{AB}^2/\delta_{AB}^2$, From the parameters listed in *Results*, we find from this relationship that $|\beta_{AB}|=6.0~{\rm cm}^{-1}$. A finite Δ_{AB} implies that the vibrational spectrum of one helix of the dimer depends on whether the other one is vibrationally excited.

The couplings established by these experiments occur between amide units that are associated with different secondary structures, and it must be a through-space electrostatic interaction. This situation is different from what usually arises with the coupling of amide-I modes that are part of the same secondary structure, because mechanical or through-bond coupling effects are almost certainly absent in the tertiary interaction. Two vibrational modes are considered to be coupled when the energy required to excite one of them depends on whether the other one is excited. This excitation sequence is precisely what is measured from the existence

of the cross-peaks in 2D IR spectroscopy. The spatial size of the dominant part of the dynamic charge distribution of the amide-I mode is small compared with separations of >4 Å between the mode centroids in different helices so that the dipole approximation to the electrostatic potential should be approximately correct (32). If consideration of the polarizability of the intervening medium is neglected, the dipolar coupling between two typical amide-I oscillators each having a transition dipole moment of 0.40 debye is 805 $\kappa(\Omega)/R^3$ cm⁻¹, where $\kappa(\Omega)$ is a geometric parameter, $-2 \le \kappa(\Omega) \le$ 2, and R is the point dipole separation in angstroms. To calibrate this interaction, the typical values for Gly₇₉ amide-I dipoles based on coordinates from the reported GpA dimer structures in other lipids (20) are an angular factor of $\kappa(\Omega) = +0.4$, a separation R =4.2 Å, and a coupling approximately +4.3 cm⁻¹. The results from FTIR and 2D IR spectroscopy interpreted by an exciton model are very consistent with values in this range.

The 2D IR spectrum contains information on the orientation of the amide groups. The ratio of the $\langle xxxx \rangle$ signal amplitude at the ¹³C=¹⁸O diagonal to that at the cross-peak is predicted (30) to be $(9\alpha_{13/18})/[(4P_2+5)\alpha_{13/16}]$, where $\alpha_{13/16}$ and $\alpha_{13/18}$ are the absorption coefficients of the ¹³C=¹⁶O and ¹³C=¹⁸O transitions, respectively $(\alpha_{13/16}/\alpha_{13/18} = 1.3)$. The observed ratio is certainly >1.4, which indicates that P_2 is negative. The ratio of the tensor quantities $\langle xxxx \rangle / \langle xyyx \rangle$ at the cross-peak is predicted from the same theory to be $(4P_2 + 5)/3P_2$, which yields an estimate of P_2 as -5/19. The observed cross-peak polarization ratio in the magnitude spectrum of 3.34 is predicted to be $(4P_2 + 5)/3|P_2|$, which yields still another estimate of a negative-valued P_2 as $P_2 = -5/14$. The latter two measurements are consistent with transition dipole angles in the range $\theta_{12} = 67-72^{\circ}$ or their supplements $\theta_{12} = 108-113^{\circ}$. The latter values are in the range for dimer structures from the coordinate sets obtained from NMR (20) but are somewhat larger than expected. The 20 NMR structures of the dimeric TM GpA (Protein Data Bank ID 1AFO) in dodecylphosphocholine micelles give the distance between the Gly₇₉/Gly₇₉ C=O groups as 3.94–5.63 Å and an expected transition dipole-dipole angle varying from 89.8° to 105.0°. The coupling between the two sites from the transition dipole interaction (33) is expected in the NMR range of +2.4 to +4.9 cm⁻¹. The dipole–dipole interaction geometric factor $\kappa(\Omega)$ is simplified to have only two angular constraints and a narrower range if there is a twofold axis compared with four when the dipoles are oriented arbitrarily. The geometric factor takes the form $\kappa(\theta_{12})$ ϕ) = 1 + (cos θ_{12} – 1) P_2 (cos ϕ), where ϕ is the azimuthal angle for one of the dipoles [the other is $(\phi + \pi)$; see Supporting Text]. For $\kappa(\theta_{12}, \phi)$ in the NMR range 0.37–0.56, the azimuth is 39° \pm 3°, again consistent with the NMR coordinates, which would place it at $35^{\circ} \pm 4^{\circ}$. This analysis locates the transition dipoles quite close to the plane perpendicular to the line joining the dipoles in the two helices. The twofold axis also interchanges the two helical axes of the dimer in the region of the interface. Because the angle that the amide-I transition dipole makes with the helix axis for a perfect α -helix is known to be $\approx 32^{\circ}$ (34), we can establish a simple relationship between (i) the crossing angle of two such dimer helices at the isotopic substitution region, (ii) the experimentally determined angle between the two transition dipoles, and (iii) the angle between the planes formed by the two transition dipoles and that by the two ideal helix axes (see Supporting Text). The latter angle is not strongly dependent on the crossing angle, so we assume that it is robust at the NMR value of -73° (20), which allows a crossing angle of 45° to be determined from the 2D IR data in SDS micelles. The coupling constants and the angles from the 2D IR spectra of GpA in SDS are consistent with a dimer interface structure that is very similar to those reported for dodecylphosphocholine (20), for which presumably the smaller crossing angle of 40° results from differences in the micelles.

The analysis of the 2D IR spectra also provides insight into the distribution of dimer structures as seen through the variance of the coupling constants. The spectral width of the ¹³C=¹⁸O



Comparison between the rephasing magnitude spectra of GpA TM heterodimer $(G_{79}^* + G_{76}^{**}) - 0.25G_{76}^{**}$ (a) and of the Ala₁₃-13C=18O-labeled, 25residue α -helix in water (10) (b). The dashed lines intersect the ¹³C=¹⁸O isotopomer peak.

monomer transition, isolated from any resonance coupling with other modes, was found to be $\approx 11.0 \pm 1.0 \text{ cm}^{-1}$ from the difference $(G_{79} + G_{79}^{**}) - 0.25G_{79}^{**}$, which displays the G_{79}/G_{79}^{**} spectrum. When the two 13C=18O modes of the homodimer G_{79}^{**}/G_{79}^{**} couple to each other, the monomer peak splits into the antisymmetric and symmetric components described above and shown in Fig. 2c, but the width of both components is expected to be decreased as a result of the excitation exchange averaging of the inhomogeneous contribution to the width. However, the width can be increased over that of the isolated mode if there is a distribution of coupling constants. In fact, the width did increase to a value of $14.0~\rm cm^{-1}$, and the detailed analysis finds a standard deviation of $2.3~\rm cm^{-1}$ in the coupling if its distribution was assumed to be Gaussian. This result confirms experimentally a contribution to the static broadening that originates directly from the distribution of dimer geometric configurations in the GpA TM domain. This distribution can be formed by variations in both $\kappa(\Omega)$ and R, which cannot be distinguished by 2D IR until information becomes available from more isotopomers. However, for illustration, if the angles are not varying much, this standard deviation corresponds to an ≈11% variation around the mean distance separating the two coupled amide units. This variance is the maximum possible in the distribution of separations of Gly₇₉ residues in the dimer. It is worth commenting that the standard deviation from the mean separation of the point dipoles obtained from the 20 reported NMR structures is 8% of the mean, so all these structures satisfy the 2D IR criteria.

More insight into the heterogeneity of the molecular environments is obtained from a comparison of the 2D IR spectra for micelle-bound GpA samples and the previously reported watersoluble helices. The 2D IR spectra of $(G_{79}^* + G_{79}^{**}) - 0.25G_{79}^{**}$ in the SDS micellar solution and the Ala₁₃- 13 C= 18 O labeled α -helix

monomer in aqueous solution (10) are shown together in Fig. 6. The difference between them is striking: The water-soluble helix (Fig. 6b) shows a much rounder peak at a lower transition frequency than the TM helix (Fig. 6a). This result implies that the distribution of amide-I frequencies is much more strongly averaged in the water-soluble helix. This averaging is due to the water dynamics, which cause fluctuations in the amide-I frequencies (35, 36). Clearly water has a greatly diminished averaging effect on the carbonyl groups in the hydrophobic core near Gly79 in the micelle. Recent 2D IR experiments on another TM peptide have provided the residue dependence of this averaging (11). The water-soluble α -helices also showed significant spectral changes occurring within the first 200 fs that were attributed to these amide frequency fluctuations, but this ultrafast dynamics was much less prominent in the current GpA TM helix samples.

Concluding Remarks

2D IR photon echo difference spectroscopy identified homodimers and heterodimers of pairs of ^{13}C — ^{16}O - and ^{13}C — ^{18}O -substituted Gly₇₉ residues on the different helices of the GpA TM dimer. The cross-peaks in the pure heterodimer 2D IR spectrum and the splitting of the homodimer peaks show that the amide-I mode is delocalized over two helices by a coupling in the range 4.3–6.3 cm $^{-1}$ that arises from through-space interactions. The geometric constraints from these experiments visualize a hydrophobic interface that is consistent with results from NMR and that sets limits on the range of geometries for dimers by specifying properties of a coupling constant distribution. These results provide a promising outlook for 2D IR applications to TM structures and their dynamical properties.

Materials and Methods

Samples. *N*-Fmoc-1-¹³C-Gly (99%), 1-¹³C-Gly (99%), ²H₂O (99%), and ¹⁸O water (95%) (Cambridge Isotope Laboratories, Andover, MA) were used to synthesize *N*-Fmoc-1-¹³C—¹⁸O-Gly (37), and a series of peptides comprising the TM region of GpA were synthesized on a 433A stepwise peptide synthesizer (Applied Biosystems, Foster City, CA) using Fmoc chemistry according to Fisher and Engelman (38), except that HATU [2-(1H-7-azabenzotriazol-1-yl)–1,1,3,3-tetramethyl uronium hexafluorophosphate methanaminium] was used as an activator and double coupling was applied only at positions 73–75, 87–88, and 90–91. The peptide sequence included residues 73–95 of human GpA and was flanked by two Lys residues at each end and amidated at the C terminus. The terminal

- 1. Hamm P, Lim M, Hochstrasser RM (1998) J Phys Chem B 102:6123-6138.
- Asplund MC, Zanni MT, Hochstrasser RM (2000) Proc Natl Acad Sci USA 97:8219– 8224.
- 3. Hamm P, Hochstrasser RM (2001) in *Ultrafast Infrared and Raman Spectroscopy*, ed Fayer MD (Dekker, New York), pp 273–347.
- 4. Mukamel S (2000) Annu Rev Phys Chem 51:691-729.
- 5. Khalil M, Demirdoven N, Tokmakoff A (2003) J Phys Chem A 107:5258-5279.
- 6. Jonas DM (2003) Annu Rev Phys Chem 54:425–463.
- 7. Woutersen S, Hamm P (2001) J Chem Phys 114:2727-2737.
- 8. Kim YS, Wang J, Hochstrasser RM (2005) *J Phys Chem B* 109:7511–7521.
- 9. Woutersen S, Hamm P (2001) J Chem Phys 115:7737-7743.
- 10. Fang C, Hochstrasser RM (2005) J Phys Chem B 109:18652-18663.
- 11. Mukherjee P, Kass I, Arkin I, Zanni MT (2006) Proc Natl Acad Sci USA 103:3528–3533.
- 12. Volkov V, Hamm P (2004) Biophys J 87:4213-4225.
- Chung HS, Khalil M, Smith AW, Ganim Z, Tokmakoff A (2005) Proc Natl Acad Sci USA 102:612–617.
- Londergan CH, Wang J, Axelsen PH, Hochstrasser RM (2006) Biophys J 90:4672–4685.
 Demirdoven N, Cheatum CM, Chung HS, Khalil M, Knoester L, Tokmakoff Δ
- Demirdoven N, Cheatum CM, Chung HS, Khalil M, Knoester J, Tokmakoff A (2004) J Am Chem Soc 126:7981–7990.
- 16. Wang J, Chen J, Hochstrasser RM (2006) J Phys Chem B 110:7545–7555.
- Fang C, Wang J, Kim YS, Charnley AK, Barber-Armstrong W, Smith AB, III, Decatur SM, Hochstrasser RM (2004) J Phys Chem B 108:10415–10427.
- 18. Popot J-L, Engelman DM (1990) Biochemistry 29:4031-4037.
- 19. MacKenzie KR (2006) Chem Rev 106:1931-1977.

Tyr₉₃ residue was mutated to Trp for improved detection. The resulting 27-residue GpA TM peptide sequence was KKITLIIFG₇₉ VMAGVIGTILLISWG₉₄IKK, in which the carbonyl of Gly₇₉ that participates intimately at the dimer helix–helix interface was selectively isotopically labeled with ^{13}C — ^{16}O or ^{13}C — ^{18}O , and another peptide with ^{13}C — ^{16}O at Gly₉₄ was prepared. The peptides were purified by reverse-phase HPLC using a protein-C4 preparative column with an i.d. of 25 mm (Grace Vydac, Hesperia, CA) and a gradient of solvents A (0.1% TFA in water) and B (0.1% TFA in 60% isopropanol/40% acetonitrile). The peptides were then lyophilized. The dry peptides were resolubilized in a 5% SDS detergent solution (45 mM sodium phosphate, uncorrected pH \approx 7.15, in 2 H₂O) at a final concentration of 5 mg/ml (1.7 mM). This procedure promotes GpA dimerization even at very high SDS concentrations (39).

2D IR Spectroscopic Method. The 2D IR measurement required ultrafast pulses, their four-wave mixing to generate the vibrational photon echo, and a heterodyned scheme to measure the generated field. Details can be found in previous reports from this laboratory (8). Briefly, an 800-nm pulse with a duration of <45-fs from a Ti:sapphire laser-amplifier system pumped a two-stage optical parametric amplifier to generate femtosecond mid-IR, 6-μm pulses, which were split into three 400-nJ, transform-limited 75-fs pulses with wavevectors \bar{k}_1 , \bar{k}_2 , and \bar{k}_3 and a bandwidth of ≈ 178 cm^{-1} . The emitted photon echo signal in the direction $-k_1 + k_2 + k_3$ was collimated with a fourth pulse that preceded it by 1.5 ps and then dispersed and recorded by an array detector as a function of the interval between incident pulses 1 and 2 (τ) and a waiting time between pulses 2 and 3 (T), which was fixed at 300 fs. The rephasing signal has pulse 1 arriving before pulse 2, whereas in nonrephasing this ordering is reversed. After double Fourier transform, the 2D correlation spectrum is the sum of the rephasing and nonrephasing spectra. The frequency step is 1.1 cm^{-1} . The center frequency of the incident pulses was chosen at the isotopomer region. The polarizations p_i of the pulses i = 1-4 are indicated explicitly in the tensor component $\langle p_1p_2p_3p_4\rangle$. All data shown in the figures are for the $\langle xxxx \rangle$ configuration.

We thank Dr. Sungwook Choi (University of Pennsylvania) for his synthesis of the ¹³C=¹⁸O-Fmoc-Gly and Profs. Nien-Hui Ge and Igor Rubtsov for valuable discussions. The research was supported by National Institutes of Health Grants GM12592 and NSFCHE (to R.M.H.) and 5P01 GM48130 (to R.M.H. and W.F.D.) and National Institutes of Health Resource RR001348.

- 20. MacKenzie KR, Prestegard JH, Engelman DM (1997) Science 276:131-133.
- 21. Javadpour MM, Eilers M, Groesbeek M, Smith SO (1999) Biophys J 77:1609–1618.
- 22. Senes A, Engel DE, DeGrado WF (2004) Curr Opin Struct Biol 14:465-479.
- Sansom MSP, Sankararamakrishnan R, Kerr ID (1995) Nat Struct Biol 2:624–631.
 Herzyk P, Hubbard RE (1995) Biophys J 69:2419–2442.
- 25. Gottschalk K-E (2005) Structure (London) 13:703–712.
- 26. Kukol A, Adams PD, Rice LM, Brunger AT, Arkin IT (1999) J Mol Biol 286:951–962.
- Sorgen PL, Hu Y, Guan L, Kaback HR, Girvin ME (2002) Proc Natl Acad Sci USA 99:14037-14040.
- Fleishman SJ, Harrington S, Friesner RA, Honig B, Ben-Tal N (2004) Biophys J 87:3448–3459.
- Manas ES, Getahun Z, Wright WW, DeGrado WF, Vanderkooi JM (2000) JAm Chem Soc 122:9883–9890.
- 30. Hochstrasser RM (2001) Chem Phys 266:273-284.
- 31. Hahn S, Lee H, Cho M (2004) J Chem Phys 121:1849-1865.
- 32. Wang J, Hochstrasser RM (2004) Chem Phys 297:195-219.
- 33. Torii H, Tasumi M (1992) J Chem Phys 96:3379-3387.
- 34. Krimm S, Bandekar J (1986) Adv Protein Chem 38:181-364.
- 35. Zanni MT, Asplund MC, Hochstrasser RM (2001) J Chem Phys 114:4579-4590.
- 36. Schmidt JR, Corcelli SA, Skinner JL (2004) *J Chem Phys* 121:8887–8896.
- 37. Torres J, Kukol A, Goodman JM, Arkin IT (2001) Biopolymers 59:396-401.
- 38. Fisher LE, Engelman DM (2001) Anal Biochem 293:102-108.
- 39. Fisher LE, Engelman DM, Sturgis JN (2003) *Biophys J* 85:3097–3105.